

Self-assembled CaF₂ nanostructures on silicon

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(Received 9 October 1998; accepted 22 February 1999)

A method for chemical imaging of CaF₂, CaF₁, and Si by scanning tunneling spectroscopy is presented. This method is utilized for identifying the growth regimes of CaF₂ and CaF₁ on stepped Si(111)7×7. For CaF₂ on Si(111), we find random islands, stripes, and ordered islands, depending on the supersaturation. For CaF₂ on a CaF₁ monolayer on Si(111), we find regular stripes that are continuous and separated from each other. CaF₂ structures are attached to the bottom edge of a step when growing directly on Si, but they prefer the top of a step edge when growing on a CaF₁ buffer layer. These highly regular, linear arrays of CaF₂ stripes and dots can serve as masks for assembling more sophisticated nanostructures. © 1999 American Vacuum Society. [S0734-2101(99)12204-8]

I. INTRODUCTION

The growth of CaF₂ on Si(111) has been studied extensively^{1–13} because this combination is lattice matched and grows epitaxially. That opens possibilities for epitaxial metal–insulator–semiconductor (MIS) devices and, eventually, to truly three-dimensional semiconductor devices. Many studies have focused on the structure and electronic states of the CaF₂/Si(111) interface which is crucial to the functioning of MIS field effect transistors (FET's). Two types of interfaces have been found, a F-terminated interface at temperatures below 700 °C, and a Ca-terminated interface from 700 °C to the desorption of CaF₂ at 800 °C (for a monolayer). The F-terminated interface has a large band gap comparable to that of bulk CaF₂ (12 eV), which is reduced to 2.4 eV at the Ca-terminated interface,¹¹ due to an extra electron in the Ca 4*s* orbital. The F-terminated interface preserves the orientation of the Si(111) substrate (type A epitaxy), while the Ca-terminated interface rotates the structure azimuthally by 180° (type B epitaxy).^{1–4}

Steps are capable of influencing the growth mode of CaF₂ in several ways. Because of the 180° rotation of the Ca-terminated interface,¹ a CaF₁ film growing on a lower terrace is mismatched with the Si on the adjacent upper terrace. Another effect has been found by Si 2*p* core level spectroscopy:⁵ A significant fraction of the interface remains F-terminated on a stepped surface, even at high annealing temperatures where the Ca-terminated structure is found on flat Si(111). Furthermore, a significant influence of steps has been inferred in detailed work on the temperature- and rate-dependent growth modes.⁷ In general, one would expect a transition from island nucleation to step flow growth with decreasing supersaturation, that is, increasing temperature, decreasing rate, and decreasing step spacing. As soon as

CaF₂ molecules are able to diffuse to the nearest step edge they become attached to it. Our goal is to observe the microscopic basis of these phenomena by using scanning tunneling microscopy (STM).

II. CHEMICALLY SENSITIVE STM

Distinguishing between CaF₂, CaF₁, and Si substrate can become quite difficult for the complex surface topography that can result from growth at stepped surfaces. To solve this problem, we have tested various spectroscopic imaging methods, such as resonant tunneling,¹⁴ and have found a new, “antiresonant” tunneling method to be very useful.¹⁵ CaF₂ and Si do not have sharp resonances that could be used for chemically-specific tunneling. However, CaF₂ exhibits a large, 12 eV band gap which suppresses the tunneling current greatly for bias voltages within the gap. That gives CaF₂ patches on the surface a dark appearance in current images taken at a bias voltage within the gap. The approach is illustrated in Fig. 1 by images of the growth pattern of islands of CaF₂ on a stepped Si(111)7×7 surface. The resolution of this chemical imaging method is at the 1 nm level or better (Fig. 1, bottom).

In practice, we stabilize the tunneling current (0.3–1.0 nA) at a sample bias voltage of +4 V, where electrons from the tip tunnel into the conduction band minimum¹³ (CBM) of CaF₂. Figure 1 (top) presents a topographic image obtained in this fashion. Sufficient sample bias voltage is essential for tunneling into insulators in general. Any attempts to stabilize the tunneling current at bias voltages below the conduction band minimum result in a tip crash, because the insulating layer acts like vacuum, thereby forcing the tip to approach the underlying Si in search of a tunneling current.

For obtaining the chemical image in Fig. 1 (center), a simultaneous current image is acquired at a bias voltage of +2 V, i.e., inside the gap of CaF₂. The topography is simplified in this image, e.g., the height difference between Si terraces is eliminated. Only the chemical composition of the surface determines the contrast, with CaF₂ dark and Si

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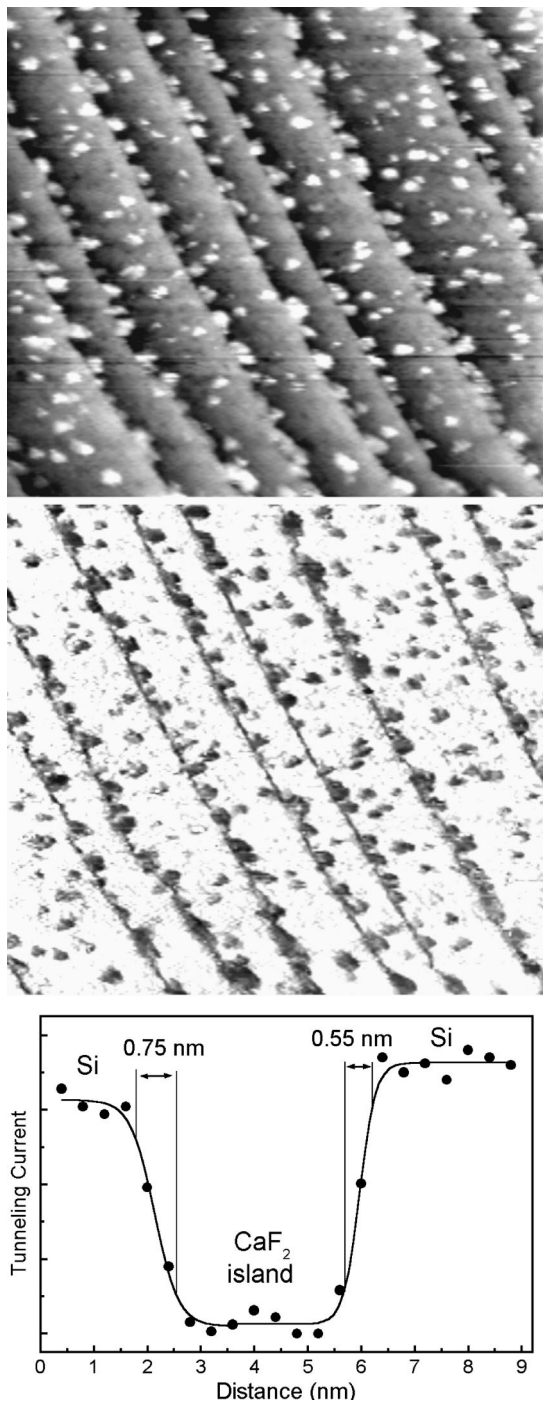


FIG. 1. Chemical imaging of small CaF_2 islands on $\text{Si}(111)7\times 7$. $100\times 80 \text{ nm}^2$. Top: Topography obtained at a sample bias of +4 V with a tunneling current of 1 nA, exhibiting Si steps with small islands of adsorbed CaF_2 . Center: Chemical image, obtained from the tunneling current distribution at a sample bias of +2 V, while stabilizing the tip height at a sample bias of +4 V. Electrons from the tip cannot enter the gap of CaF_2 , resulting in a low current for CaF_2 patches (dark areas). Height difference between terraces is eliminated and the contrast is reversed, with Si appearing bright and CaF_2 dark. Bottom: Line scan across a CaF_2 island demonstrating a resolution of better than 1 nm. This resolution limit corresponds to the change in the tunneling current from 85% to 15% of the maximum value.

bright. For CaF_2 islands on a Si substrate, the assignment of different features is not very ambiguous. However, in case of a more complicated surface structure, such as bunched Si steps, step flow, and step etching, the difference between the

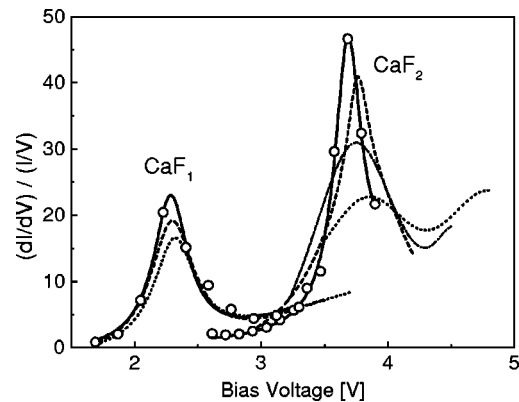


FIG. 2. Scanning tunneling spectra of CaF_1 and CaF_2 showing the onset of the conduction band minimum at different voltages. Different line types represent different tip heights.

substrate and the adsorbate is not always apparent in the topography image. The current image also allows identification of the chemical state (CaF_2 vs CaF_1) due to the difference in band gap (more on that in Fig. 2).

It is well known that a mixture of topographic and chemical contrast is seen in the topography. For example, at a sample bias of +4 V (Fig. 3) the CaF_2 islands and stripes appear higher (brighter) than equivalent Si terraces, even though CaF_2 is lattice matched to Si within <1%. There must be enhanced tunneling into states near the conduction band minimum of CaF_2 , which is further supported by the resonance in the tunneling spectra at a bias voltage close to the CBM (Fig. 2).

The spectroscopic underpinnings of our chemical imaging method are shown in Fig. 2. Essentially, we take advantage of an unexpectedly large jump in the conductivity at the conduction band minimum of insulators, which allows us to use the energy of the CBM relative to the Fermi level as the chemical identifier. The differential conductance (dI/dV) or its normalized version $(dI/dV)/(I/V)$ can be obtained from the scanning tunneling spectroscopy $I(V)$ curves. The conduction band edges correspond to onsets in the (dI/dV) curves and become well-defined peaks in the normalized spectra (Fig. 2). While the exact origin of such strong and narrow peaks is still being investigated,¹⁵ it is interesting to note that the 3.7 eV peak in Fig. 2 for CaF_2 is consistent with the 3.6–3.8 eV CBM position expected from macroscopic observation techniques,^{5,6} but the 2.3 eV CaF_1 peak is higher than the expected¹¹ 1.4–1.8 eV.

III. GROWTH MODES AT STEPS

General growth models^{7,16} predict a transition from island nucleation to step flow growth with increasing temperature, decreasing rate, and decreasing step spacing. Growth in general is a nonequilibrium process for which a particular outcome is determined by the combined influence of these parameters, together with the total coverage. The growth

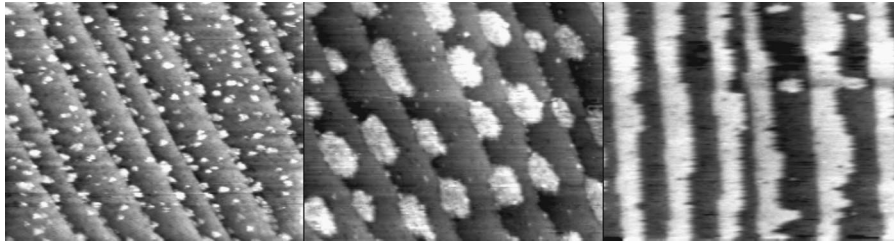


FIG. 3. Three growth regimes for CaF₂ on Si(111)7×7 with single steps, tilted 1.1° towards $(\bar{1} \bar{1} 2)$. Downhill is to the right. CaF₂ appears bright. 100×80 nm². Left: Island growth at 550 °C (compare Fig. 1). The island size (≈ 2 nm) decreases at lower temperature in proportion to the shrinking diffusion length. Center: Strings of regular islands (10 nm) attached to the Si step edges. These are formed at higher mobility and deposition rate (growth at 650 °C and post-annealed at 700 °C for 5 min). Right: Step flow growth at 600 °C, producing CaF₂ stripes 5–10 nm wide.

modes studied for CaF₂ in this work can be classified according to the respective temperature–rate–coverage combinations:

(1) Low temperature¹⁷ (550 °C), low deposition rate, and low coverage result in randomly distributed small islands (Fig. 3, left). The average size and separation are determined by the CaF₂ diffusion length.

(2) Increasing temperature (650 °C) and deposition rate, together with a low coverage result in strings of CaF₂ “dots” attached to the Si step edges (Fig. 3, center). Under these conditions, CaF₂ molecules are able to diffuse towards the preferred sites adjacent to Si steps and form larger dots. The dot size and spacing appear to be self-limited,¹⁸ leading to uniform structures.

(3) Moderate increase of temperature (600 °C) and coverage, combined with lower deposition rate result in typical step-flow growth (Fig. 3, right).

(4) Substantial increase in temperature (700 °C and above) leads to desorption processes. On larger terraces, a conversion of CaF₂ to CaF₁ interface is observed. On narrow terraces, we find that CaF₂ desorbs by etching away Si step edges, leading to Si step edges receding between remaining CaF₂ stripes and dots. A likely reaction is the formation of volatile SiF₄ plus a remaining Ca layer on Si, which we see as single-domain 3×1 structure.¹⁹

(5) Moderate temperature (610–630 °C) combined with a coverage between 1 and 2 monolayers lead to a qualitatively different growth mode where highly perfect CaF₂ stripes are formed on top of a CaF₁ monolayer (Fig. 4, top).

At the lowest growth temperatures, islands nucleate spontaneously and ripen by sweeping up CaF₂ molecules that land within a diffusion length of an island. Consequently, the size of the islands and their average spacings are related to the diffusion length.¹⁶ For growth at 550 °C and 5 s (Fig. 3, left), we find an average island diameter of about 2 nm, for growth at 650 °C and 10 s, the average island diameter is about 10 nm. Islands are noticeably absent in the region adjacent to the upper edge of a step. A systematic analysis of images taken in the island growth regime at 550 °C verifies this observation. It suggests that CaF₂ is bound less strongly at the upper edge of a step.

At higher temperatures and faster deposition, CaF₂ forms a regular array of islands attached to the step edges. In Fig. 3

(center) the coverage is 1/5 of a monolayer of CaF₂ deposited at 650 °C with a brief postanneal at 700 °C. This growth mode is reminiscent of the Stranski–Krastanov growth in two dimensions, except that the CaF₂ islands are much more regular.²⁰ A competition between ripening into large dots and an upper size limit imposed by misfit strain may be responsible for size homogeneity. High deposition rate and

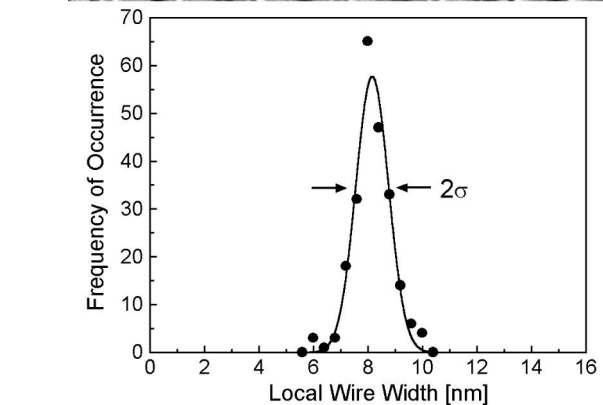
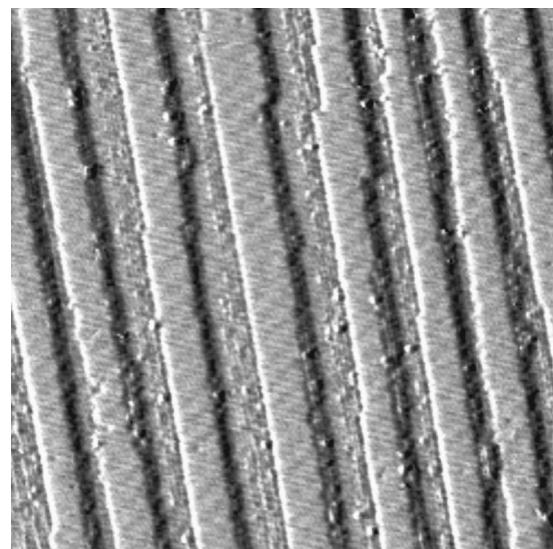


FIG. 4. CaF₂ stripes grown on a Si(111) surface coated by CaF₁. Top: STM image, showing the x derivative of the tip height, simulating illumination from the left. Note that the CaF₂ stripes are attached to the top of the step edges. Downhill is to the right. 100×100 nm². Bottom: Width distribution along one of the stripes in Fig. 4 top.

increased mobility of CaF₂ at this elevated temperature lead to a supersaturation nucleation and growth when the formation of new nucleation sites is suppressed after a critical island concentration is attached. Nucleation at step edges is preferred due to the higher coordination of such sites.

The Si step separation of 15 nm and the dots sizes of 7×10 nm in growth mode 2 lead to arrays of dots that may be used as precursors for creating ultradense storage media. Their density is on the scale of 10¹¹ dots/cm², or Teradots per square inch, in units commonly used in the storage industry. This represents a thousand-fold increase compared to commercially available densities of few Gigabits per square inch in hard disks. The linear arrangement of the islands along the straight step edges makes this growth mode particularly suited for possible linear readout architectures, such as shift registers or scanning probes and reading heads.

The main difference between modes shown in Fig. 3 center and right is that in the latter case deposition rate is lower which combined with considerable mobility of CaF₂ molecules at 600 °C gives rise to the step flow growth mode. Higher coverage allows the formation of CaF₂ stripes which decorate the Si(111) step edges (Fig. 3, right). For growth at 600 °C in 10 s and half a monolayer coverage, we find that on the terraces narrower than 17 nm all the CaF₂ has been swept into the stripes at the step edge. Given approximately half-monolayer coverage molecules had to diffuse at least across half the terrace (8–9 nm) to become attached. The first islands remaining on the clean part of terraces appear for terrace widths of about 20–25 nm, suggesting that CaF₂ molecules could not diffuse farther than 10–12 nm. The two observations combined lead to a diffusion length of the order of 10 nm at these growth conditions.

Figure 4 (top) presents an example of a more complicated growth mode. After depositing 1.5 monolayers of CaF₂ at 610–630 °C and a postanneal to 830 °C for 30 s, a complete overcoat of CaF₁ is established with CaF₂ stripes on top of it. Chemical imaging confirms the identity of the two species. Unlike the CaF₂ on Si stripes and dots described above, the stripes grown on CaF₁ are formed exclusively on the top edges of Si steps.²⁰ In fact they avoid the bottom edge to such extent, that even for coverage very close to 2 monolayers, there is always an uninterrupted CaF₁ gap between a stripe and a bottom edge of a Si step. A simple explanation for this strikingly different behavior comes from the fact that the Ca-terminated interface (i.e., CaF₁ layer underneath CaF₂) rotates the lattice by 180° around the normal.⁴ The CaF₂ stripes are then laterally mismatched with the adjacent Si step and avoid it. Position on top of the step on the other hand allows CaF₂ to maintain its preferred orientation. Note that the F-terminated interface does not induce such a 180° rotation,⁴ so the CaF₂ structures grown at lower temperature and coverage are matched with the adjacent Si steps and preferentially grow attached to their bottom edges (Fig. 3).

The interaction described above leads to formation of continuous and completely separated CaF₂ stripes²⁰ (Fig. 4, top) and also helps to keep them rather homogeneous. With the help of a pattern recognition algorithm we can quantify

the width distribution along a single stripe. The resulting distribution is presented in Fig. 4 (bottom). A simple Gaussian fit (solid curve) gives a mean width of 8 with 0.6 nm standard deviation. Such homogeneous and continuous stripes can serve as templates for selective deposition of insulated metallic nanowires with widths in the single digit nanometer range. Some possible chemically selective deposition processes are being currently investigated.²¹

IV. SUMMARY AND OUTLOOK

In summary, we have developed a chemical imaging method for STM, using antiresonant tunneling into the band gap of insulating CaF₂. This technique should be generally applicable to the chemical imaging of insulators with different band gap by STM. Using it for determining the microscopic growth mode of CaF₂ on stepped Si(111)7×7, we have found interesting effects, such as CaF₂ stripes in the step flow regime, regular CaF₂ island arrays at low coverage and high supersaturation, the repulsion of CaF₂ on Si from the upper step edge, and repulsion of CaF₂ on CaF₁ from the lower step edge induced by the Ca-terminated interface lattice rotation.

A longer range goal of this work is establishing universal methods for nanolithography or wires by self-assembly at step edges.²² It is already possible to produce highly perfect step arrays²³ on Si(111)7×7. Stripes and dots of CaF₂ attached to Si step edges should be able to play a role similar to the photoresist in ordinary microlithography, passivating the underlying Si. That would provide the essential ingredients for controlled nanolithography of wires and dots.

ACKNOWLEDGMENTS

This work was supported by NSF under Award Nos. DMR-9624753 and DMR-9632527.

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